# Synthesis and Magnetic Properties of (Zn<sup>2+</sup>Ti<sup>4+</sup>) Substituted W-type and Y-type Ferrites

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We report the synthesis of  $(Zn^{2+}Ti^{4+})$  substituted hexagonal ferrites,  $BaZn_2(Zn_{0.5}Ti_{0.5})_xFe_{16-x}O_{27}$  (W-type) and  $Ba_2Zn_2(Zn_{0.5}Ti_{0.5})_xFe_{12-x}O_{22}$  (Y-type), and the relationship between their composition and magnetic properties. Substituted W-type ferrite was synthesized in the region of  $0.0 \le x \le 1.5$ . A decrease in the Curie temperature of the W-type phase indicated that the exchange interactions between the Fe<sup>3+</sup> ions weakened due to the substitution of Zn<sup>2+</sup> and Ti<sup>4+</sup> ions. Spontaneous magnetization ( $\sigma_s$ ) at x = 0.5 was similar to that at x = 0.0. This reflects the equal distribution of nonmagnetic Zn<sup>2+</sup> and Ti<sup>4+</sup> ions to spin down sites and spin up sites, respectively. At x > 0.5,  $\sigma_s$  decreased clearly, which was possibly caused by a deviation from the collinear arrangement. Substituted Y-type ferrite was also obtained up to x = 0.5. The Y-type phase is more unstable than the W-type phase with respect to this substitution rate.

Key words: W-type ferrite, Y-type ferrite, (ZnTi) substitution, solid state reaction, X-ray diffraction, Curie temperature, spontaneous magnetization, cation distribution

#### 1. Introduction

The hexagonal ferrites have been investigated for electronic devices at microwave frequencies since their discovery in 1950s <sup>1)</sup>. Their availability at high frequencies is caused by the large magnetic anisotropy constants and the preference to several magnetized directions, parallel or perpendicular to the hexagonal c-axis. Most of ferrites also have low eddy current loss at high frequency and corrosive resistance because they are oxides.

The crystal structure of hexagonal ferrites is based upon the close-packed structure of large oxygen anions and barium cations. The small transition metal cations (Fe<sup>3+</sup> and Me<sup>2+</sup>) are located between the larger ions. The structures of hexagonal ferrites can be represented by the stacking combinations along their hexagonal c-axis of R, S, and T blocks 2). The chemical formulae for these blocks are as follows:  $R = (BaFe_6O_{11})^{2^\circ}$ ,  $S = Me_2Fe_4O_8$  /  $(Fe_6O_8)^{2+}$ , T = Ba<sub>2</sub>Fe<sub>8</sub>O<sub>14</sub>. The composition of the S block corresponds with the two formula units of the spinel ferrite. The major hexagonal ferrites are described by the following symbols: RSR\*S\* for M-type (BaFe<sub>12</sub>O<sub>19</sub>),  $(RSR^*S^*S^*)_3$  for X-type  $(Ba_2Me_2Fe_{28}O_{46})$ ,  $RSSR^*S^*S^*$  for  $(BaMe_2Fe_{16}O_{27}),$ W-type  $(TS)_3$ for Y-type  $(Ba_2Me_2Fe_{12}O_{22}),$ RSTSR\*S\*T\*S\* for Z-type  $(Ba_3Me_2Fe_{24}O_{41}),$  $(RSTSR^*S^*)_3$ and for U-type (Ba<sub>4</sub>Me<sub>2</sub>Fe<sub>36</sub>O<sub>60</sub>) <sup>3), 4)</sup>. The superscript \* denotes that the layer is turned for 180° around the *c*-axis.

M-type barium ferrite (BaFe<sub>12</sub>O<sub>19</sub>) is widely used for permanent magnets due to its large magneto-crystalline anisotropy <sup>5)</sup>. In addition, many research groups have investigated barium ferrite as microwave absorbers <sup>6)-8)</sup> and magnetic recording media <sup>9)</sup> by modifying its anisotropy. The control of its magnetic property is achieved by the replacement of Fe<sup>3+</sup> ions by Al<sup>3+</sup> 10)-13),  $Cr^{3+}$  $^{13)\cdot15)}$  ions or pairs of  $(Me^{2+}Mt^{4+})$  ions such as  $(Co^{2+}Ti^{4+})$  $^{16,}$  $^{17)},$   $(Zn^{2+}Ti^{4+})$  $^{18),}$  $^{19)},$   $(Mn^{2+}Ti^{4+})$  $^{6),}$  $^{20)},$   $(Ni^{2+}Sn^{4+})$  $^{21)},$   $(Co^{2+}Sn^{4+})$  $^{22)},$   $(Co^{2+}Zr^{4+})$  $^{23),}$  $^{24)},$  where the Me and Mt represent divalent and tetravalent cations, respectively. The substitution effect on anisotropy is so significant that even the hard magnetic property of BaFe1\_2O\_{19} was changed to soft magnetic by substitution of  $(Co^{2+}Zn^{2+})Ti^{4+}$  for Fe^{3+}25).

On the other hand, W-type and Y-type ferrites have different magnetic properties from those of M-type ferrite due to their different stacking structures. W-type ferrite can possess higher saturation magnetization than that of M-type ferrite <sup>26)</sup>. Y-type ferrite has planar anisotropy <sup>27)</sup>, which gives hope for high-frequency application. Therefore, the control of the magnetic properties of W-type and Y-type ferrites has been an important issue.

To the best of our knowledge, however, there are few investigations on the substituted W-type and Y-type ferrite. Maeda *et al.* reported the preparation by a conventional ceramic technique, magnetic properties, and microwave absorption properties of the  $(Mn^{2+}Sn^{4+})$ and  $(Mn^{2+}Zr^{4+})$  substituted W-type ferrite <sup>28)</sup>. Ul-ain *et al.* reported the synthesis by sol-gel method and catalytic activities of the  $(Mn^{2+}Ti^{4+})$  substituted Y-type ferrite <sup>29)</sup>. Also, substitution limits of Me<sup>2+</sup>Mt<sup>4+</sup> for Fe<sup>3+</sup> in W-type and Y-type structures are unclear in contrast with that in the M-type structure where complete substitution of Co<sup>2+</sup>Ti<sup>4+</sup> for Fe<sup>3+</sup> is possible to form BaCo<sub>6</sub>Ti<sub>6</sub>O<sub>19</sub> <sup>30)</sup>.

We report the synthesis of the  $(Zn^{2+}Ti^{4+})$  substituted hexagonal ferrites,  $BaZn_2(Zn_{0.5}Ti_{0.5})_xFe_{16}$ ,  $O_{27}$  (W-type) and  $Ba_2Zn_2(Zn_{0.5}Ti_{0.5})_xFe_{12}$ ,  $O_{22}$  (Y-type), and the relation between their composition and magnetic properties in this paper. Here, we employed  $Zn^{2+}Ti^{4+}$  cations for substitution because an increase of the magnetization was expected if nonmagnetic cations replaced Fe cations in spin down sublattices as observed in substituted spinels <sup>31)</sup>.

## 2. Experimental procedure

The samples were prepared by solid-state reaction. We used BaCO<sub>3</sub>, ZnO, TiO<sub>2</sub>, and  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> as starting materials. They were mixed in the desired composition of Ba : Zn : (Zn<sub>0.5</sub>Ti<sub>0.5</sub>) : Fe = 1 : 2 : x : (16-x) for the W-type ferrite and Ba : Zn : (Zn<sub>0.5</sub>Ti<sub>0.5</sub>) : Fe = 2 : 2 : x : (12-x) for the Y-type ferrite with  $0.0 \le x \le 2.0$  in a ball-milling pot for 24 h. The mixed powders were pressed into disks and then pre-sintered at 900°C in air for 5 h. The products were ground in a mortar and then pulverized for 10 min. at the rate of 1100 r.p.m. in a vario-planetary mill. The fine powders were pressed into disks again and then sintered at 1100-1350°C in air for 5 h. Finally, the products were ground into powders in a mortar.

The crystal structures were examined by powder X-ray diffraction (XRD) with Cu-K $\alpha$  radiation. The magnetizations were measured with a vibrating sample magnetometer (VSM), and a superconducting quantum interference device magnetometer (SQUID).

## 3. Results and discussion

We identified the crystalline phases of the W-type composition samples with  $0.0 \le x \le 2.0$  sintered at 1200-1350°C by XRD. Fig. 1 shows phase relationship between sintering temperature and substitution rate *x*. At  $0.0 \le x \le 1.5$ , the W-type phase formed in the samples



**Fig. 1** Relationship between obtained crystalline phase and sintering temperature and substitution rate x for samples whose raw materials were mixed in composition of Ba :  $Zn : (Zn_{0.5}Ti_{0.5}) : Fe = 1 : 2 : x : (16 \cdot x)$  with  $0.0 \le x \le 2.0$ . Phases in parentheses are minority phases.

sintered at 1250°C, provided that the W-type phase was the minority phase for the sample with x = 1.5. The samples with  $0.0 \le x \le 1.5$  sintered at 1300°C and 1350°C consisted of the main W-type phase. On the other hand, at sintering temperature of 1350°C, the sample contained the M-type and spinel minority phases. Since this temperature is near to the melting point of the other barium-iron oxides (e. g. Ba<sub>2</sub>Fe<sub>2</sub>O<sub>5</sub>;  $T_m = 1370$ °C <sup>32)</sup>), the W-type phase is likely to decompose. These results show that the sintering temperature of 1300°C is the suitable condition for the synthesis of the substituted W-type ferrites.

Fig. 2 shows XRD patterns of W-type composition samples with Ba : Zn :  $(Zn_{0.5}Ti_{0.5})$  : Fe = 1 : 2 : x : (16-x) sintered at 1300°C. The samples had the W-type structure at  $0.0 \le x \le 1.5$ . At x = 1.5, there were some



**Fig. 2** XRD patterns for products sintered at 1300°C whose raw materials were mixed in composition of Ba :  $Zn : (Zn_{0.5}Ti_{0.5}) : Fe = 1 : 2 : x : (16 \cdot x)$  with  $0.0 \le x \le 2.0$ .



**Fig. 3** XRD patterns for products sintered at 1100°C whose raw materials were mixed in composition of Ba : Zn :  $(Zn_{0.5}Ti_{0.5})$  : Fe = 2 : 2 : x :  $(12 \cdot x)$  with  $0.0 \le x \le 2.0$ .

weak diffraction peaks of the M-type and spinel ferrites. On the contrary, the peaks of the W-type phase disappeared in the pattern at x = 2.0. It should be noted that the first and second peaks of the W-type phase do not overlap those of the spinel and M-type phases as comparing the patterns between  $x \le 1.5$  and x = 2.0. Fig. 3 shows XRD patterns of Y-type composition samples with Ba :  $Zn : (Zn_{0.5}Ti_{0.5}) : Fe = 2 : 2 : x : (12-x)$  sintered at 1100°C. We obtained the Y-type phase in the samples with x = 0.0 and 0.5. On the other hand, the diffraction peaks of the Y-type phase weakened at  $x \ge 1.0$  and finally disappeared at x = 2.0. The phase identification was based on the diagrams of the BaO-ZnO-Fe<sub>2</sub>O<sub>3</sub> 33), BaO-ZnO-TiO<sub>2</sub> <sup>34)</sup> and BaO-Fe<sub>2</sub>O<sub>3</sub>-TiO<sub>2</sub> <sup>35)</sup> systems, which showed no corresponding compounds except for the spinel and Y-type phases. The maximum substitution rate of (ZnTi)/(Fe+(ZnTi)) to form the single Y-type phase  $(0.5/12 \sim 4.2\%)$  is smaller than that to form the single W-type phase  $(1.0/16 \sim 6.3\%)$ . Therefore, the Y-type phase is less stable than the W-type phase for (Zn<sup>2+</sup>Ti<sup>4+</sup>) substitution.



**Fig. 4** Temperature dependences of magnetization at  $\mu_0 H = 0.1$  T for samples sintered at 1300°C whose raw materials were mixed in composition of BaZn<sub>2</sub>(Zn<sub>0.5</sub>Ti<sub>0.5</sub>)<sub>x</sub>Fe<sub>16·x</sub>O<sub>27</sub> with 0.0  $\leq x \leq$  1.5. Inset shows Curie temperatures *T*<sub>C</sub> of those samples.

Fig. 4 shows temperature dependences of magnetization at  $\mu_0 H = 0.1$  T for W-type composition samples of BaZn<sub>2</sub>(Zn<sub>0.5</sub>Ti<sub>0.5</sub>)<sub>x</sub>Fe<sub>16·x</sub>O<sub>27</sub> (0.0  $\leq x \leq 1.5$ ) sintered at 1300°C. Inset of Fig. 4 shows the Curie temperature  $T_{\rm C}$  of the W-type phase as a function of the composition x, where  $T_{\rm C}$  was estimated by linear extrapolation to zero-magnetization from the inflection point of the M-T curve. The Curie temperature  $T_{\rm C}$  of the no-substitution sample (x = 0.0) was 305°C, which is in agreement with the previous report about BaZn<sub>2</sub>Fe<sub>16</sub>O<sub>27</sub> single crystals  $^{36)}$ . T<sub>c</sub> linearly decreased with increasing x at  $x \leq 1.5$ , suggesting the exchange interactions between Fe<sup>3+</sup> ions weakened. This fact implies that  $Zn^{2+}$ and Ti<sup>4+</sup> cations successfully replaced Fe<sup>3+</sup> cations in the W-type structure at  $x \le 1.5$ .

Fig. 5 shows magnetization curves at T = 5 K (=  $-268^{\circ}$ C) for W-type samples of BaZn<sub>2</sub>(Zn<sub>0.5</sub>Ti<sub>0.5</sub>)<sub>x</sub>Fe<sub>16·x</sub>O<sub>27</sub> ( $0.0 \le x \le 1.5$ ) sintered at 1300°C. The magnetization curve of the substituted samples at x = 0.5 was almost the same as that of the no-substitution sample (x = 0.0). Then, the magnetization decreased with increasing the substitution rate from x = 0.5 up to 1.5. This result is consistent with the fact that M-type ferrite showed similar Zn-Ti substitution dependence of spontaneous magnetization  $^{37}$ , where the magnetization began to



**Fig. 5** Magnetization curves at T = 5 K (= -268°C) for samples sintered at 1300°C whose raw materials were mixed in composition of BaZn<sub>2</sub>(Zn<sub>0.5</sub>Ti<sub>0.5</sub>)<sub>x</sub>Fe<sub>16</sub>·<sub>x</sub>O<sub>27</sub> with  $0.0 \le x \le 1.5$ .



**Fig. 6** Spontaneous magnetization and high field differential susceptibility for samples sintered at 1300°C whose raw materials were mixed in composition of  $BaZn_2(Zn_{0.5}Ti_{0.5})_xFe_{16\cdot x}O_{27}$  with  $0.0 \le x \le 1.5$ .

decrease above a certain Zn-Ti substitution rate.

Fig. 6 shows spontaneous magnetization  $\sigma_{\rm S}$  and high field differential susceptibility  $\chi_{\rm d}$  as a function of the composition x. They were estimated by linear extrapolation of the magnetization curves from a high field region. The spontaneous magnetization  $\sigma_{\rm S}$  slightly decreased from x = 0.0 to 0.5. In the range of 0.5 <  $x \le 1.5$ , the decrease of  $\sigma_{\rm S}$  became more obvious. The substitution effect on  $\sigma_{\rm S}$  seems to reflect  ${\rm Zn}^{2+}$ -Ti<sup>4+</sup> cation distribution in the W-type structure.

In order to discuss the relationship between the change in  $\sigma_{\rm S}$  and the cation distribution in the W-type structure, we consider the spin configuration of the W-type structure. Table 1 shows the distribution of the transition metal sites and the expected directions of the magnetic moments in the R and S blocks of the W-type structure <sup>38)</sup>. Fig. 7 also shows magnetic structure of W-type ferrite, corresponding to Table 1.

Concerning the cation distribution in the W-type structure, we assume that the  $Zn^{2+}$  ions prefer the 4eand  $4f_{\rm IV}$  sites and that Ti<sup>4+</sup> ions prefer the 12k sites. In the case of the (ZnTi) substituted M-type ferrite <sup>37)</sup>, Zn<sup>2+</sup> ions tend to enter the tetrahedral 4 fiv sites in the S block. Also,  ${\rm Ti}^{4+}$  ions tend to enter the octahedral  $4{\it f}_{\rm VI}$  (the R block) sites and the 12k (the S and R blocks border) sites. In our W-type ferrite case, even with no substitution sample of x = 0.0, however, the positively charged part of two S blocks of (Zn<sub>2</sub>Fe<sub>10</sub>O<sub>16</sub>)<sup>2+</sup> and the negatively charged R block of (BaFe<sub>6</sub>O<sub>11</sub>)<sup>2-</sup> are alternately stacked up as  $SSRS^*S^*R^*$ . The additional  $Zn^{2+}$  ions seem to occupy 4eand 4 fiv sites in the S blocks like the ZnTi-doped M-type ferrite. In order to keep the local electronic balance in the additional Zn<sup>2+</sup> doped S blocks, Ti<sup>4+</sup> ions may prefer the 12k sites to the  $4f_{\rm VI}$  sites because of the proximity to the  $Zn^{2+}$  sites in the S block.

We discuss the effect of the nonmagnetic ions upon the spontaneous magnetization  $\sigma_{\rm S}$  depending on x. Between x = 0.0 and 0.5, the occupation of spin down (4e and 4fiv) sites by Zn<sup>2+</sup> ions can be cancelled out by the occupation of spin up (12k) sites by Ti<sup>4+</sup> ions. Therefore, the decrease of  $\sigma_s$  is slight in this range. At x > 0.5, the obvious decrease of  $\sigma_{\rm S}$  is possibly caused by the deviation from the collinear arrangement which can be dominant at high substitution rates of Zn<sup>2+</sup> and Ti<sup>4+</sup> ions <sup>37)</sup>. To our estimation, further ascertain experimental investigations are needed to estimate the spin configuration and the cation distribution by Rietveld analysis on powder neutron diffraction 39) and X-ray absorption fine structure analysis <sup>40</sup>.

The high field differential susceptibility  $\chi_d$  for the substituted W-type ferrite increases with increasing x as shown in Fig. 6. It suggests that the magnetic structure of the substituted W-type ferrite becomes non-collinear especially at x > 0.5. It is consistent with the interpretation of the variation in  $\sigma_s$  discussed above.

**Table 1** Coordination, number of ions, and spinorientation for various cation sublattices of W-typeferrite.

Block	Sublattice	Coordination	Number per block	Spin
S-S	6g	Octahedral	3	Up
$\mathbf{S}$	4e	Tetrahedral	2	Down
S	4f	Octahedral	2	Up
S	$4f_{ m IV}$	Tetrahedral	2	Down
S-R	12k	Octahedral	6	Up
R	$4f_{ m VI}$	Octahedral	2	Down
R	2d	Trigonal -bipyramidal	1	Up



Fig. 7 Magnetic structure in half of unit cell (in formula unit ) of W-type ferrite.

## 4. Conclusions

We report the synthesis condition and the magnetic properties of the  $(Zn^{2+}Ti^{4+})$ -substituted W-type and Y-type hexagonal ferrites that have hardly been investigated so far. The synthesis condition shows the substitution limits of  $Zn^{2+}Ti^{4+}$  for Fe<sup>3+</sup> in the W-type and Y-type ferrites, suggesting the stability of the W-type and Y-type structures.

The W-type composition samples sintered at 1300°C had the W-type structure in the region of  $0.0 \le x \le 1.5$ . The Curie temperature of the W-type phase decreased linearly with increasing x at  $x \le 1.5$ . It suggests that the exchange interactions between the Fe<sup>3+</sup> ions weakened due to the substitution of nonmagnetic Zn<sup>2+</sup> and Ti<sup>4+</sup> ions in the W-type structure. Spontaneous magnetization  $\sigma_{\rm S}$  at x = 0.5 was similar to that at x = 0.0. It reflects the equal distribution of nonmagnetic  $Zn^{2+}$  and  $Ti^{4+}$  ions to spin down sites and spin up sites, respectively. At x > 0.5,  $\sigma$ s decreased clearly, which is possibly caused by the deviation from the collinear arrangement. It is consistent with the increase in the high field differential susceptibility  $\chi_d$  with x.

The Y-type single phase was also obtained for the Y-type composition samples up to x = 0.5. The Y-type phase is more unstable than the W-type phase with respect to this substitution rate.

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